

# Download Stoichiometry Worksheet 7

7. How many moles of O<sub>2</sub> are produced when 1.26 moles of H<sub>2</sub>O is reacted? ... may moles of O<sub>2</sub> must have been made? (a bit challenging, but just think about it and you can probably figure it out) Stoichiometry Worksheet and Key 1.65 mol KClO<sub>3</sub> mol KClO<sub>3</sub> mol O<sub>2</sub> = mol O<sub>2</sub> ... stoichiometry\_1\_worksheet\_and\_key.docx Stoichiometry 1. Showing top 8 worksheets in the category - Stoichiometry 1. Some of the worksheets displayed are Stoichiometry practice work, Stoichiometry practice work, Stoichiometry 1 work and key, Stoichiometry work 1, Stoichiometry problems 1 work, Stoichiometry work 1 answers, Stoichiometry problem 1, Chapter 6 balancing stoich work and key.5. Balancing and Stoichiometry: a. H<sub>2</sub> + Cl<sub>2</sub> ® HCl (needs balanced) How many grams of HCl can be produced if 7.25 g of Cl<sub>2</sub> is reacted with an unlimited supply of H<sub>2</sub>? b. Al + Fe<sub>2</sub>O<sub>3</sub> ® Al<sub>2</sub>O<sub>3</sub> + Fe (needs balanced) How many grams of Fe can be produced when 10.0g of Al is reacted with an excess (unlimited) supply4. Given the following equation: Na<sub>2</sub>O + H<sub>2</sub>O ---> 2 NaOH How many grams of NaOH is produced from 1.20 x 10<sup>2</sup> grams of Na<sub>2</sub>O? How many grams of Na<sub>2</sub>O are required to produce 1.60 x 10<sup>2</sup> grams of NaOH? 5., Stoichiometry Worksheet 7.

## Other Files :

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