

Download Practice Stoichiometry

Determine the amount (in moles) of a product from a given amount of one reactant. Practice Problems: Stoichiometry. Balance the following chemical reactions: Hint a. $\text{CO} + \text{O}_2 \rightarrow \text{CO}_2$ b. $\text{KNO}_3 \rightarrow \text{KNO}_2 + \text{O}_2$ c. $\text{O}_3 \rightarrow \text{O}_2$ d. $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \text{H}_2\text{O}$ e. $\text{CH}_3\text{NH}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{N}_2$ Hint f. $\text{Cr}(\text{OH})_3 + \text{HClO}_4 \rightarrow \text{Cr}(\text{ClO}_4)_3 + \text{H}_2\text{O}$ Write the balanced chemical equations of each reaction: Questions pertaining to stoichiometry If you're behind a web filter, please make sure that the domains *.kastatic.org and *.kasandbox.org are unblocked. Stoichiometry Problems and Practice Stoichiometry problems are one of the most difficult areas in general chemistry. The first step is to master the basics—that's what this section is about. To build your stoichiometry skills you'll get the basic information and examples, lots of practice with support, and then a quiz to make sure you've got it., Practice Stoichiometry .

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